<u>Lab #2 – Observation Explosion!</u>

For Every Lab:

- Read all instructions before beginning to work.
- Follow the instructions carefully.
- Work as a team with your lab partner or group.
- Record all observations as you do each step.
- Make sketches where appropriate.
- Clean up your lab table and sink top before the end of the period.

Materials:

* 250 ml beaker	* stirrer (plastic teaspoon)	*balance
250 IIII beakei	Stiffer (blastic teaspoor)	balance

* 100 ml graduated cylinder * Baking Soda * White vinegar

Procedure:

- 1. Add 10g of baking soda to the empty 250 ml beaker.
 - a. (Return Baking Soda Spoon to jar <u>DO NOT stir with it!</u>).
- 2. Use the appropriate size graduated cylinder(s) to measure 120 ml of tap water.
 - a. Pour the water **SLOWLY** into the beaker and stir gently with the spoon.
 - ь. In the observations section, <u>describe</u> the appearance of your solution.
 - c. <u>Make & label sketches</u> of the beaker during each step of your experiment (in observations section)
- 3. Measure 30 ml of vinegar with a graduated cylinder.
 - a. Add the vinegar to the solution in the beaker <u>VERY SLOWLY without</u> <u>stirring!</u>

- b. Has the appearance of your solution changed? <u>Describe it (in observations</u> section). What happened?
- 4. Carefully put three or four items of your choice (any combination, from the front table) into the solution and watch what happens. Look for details and not just the obvious things to happen. You may need to watch carefully for more than 5 minutes, so please be patient. What did you observe? Describe it (in observations section).

Name	Date	Per	

Lab #2- Observation Explosion!

Purpose: Why do you think we're doing this lab?

Hypothesis: (read all instructions, then form your "Educated Guess") What will occur when baking soda, vinegar, water, and the items are combined?

 If ________, then

 _______, because

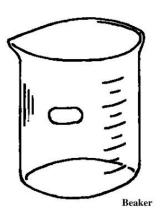
_____•

Observations: (3 Labeled Sketches – one each from steps 2, 3, and 4)

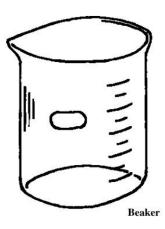
Labeled sketches:

Written observations:

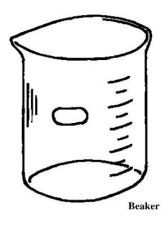
(3 detailed descriptions per sketch - ok to bullet)



Step 2.



Step 3.



Step 4.

<u>Analysis Questions</u>: – use COMPLETE SENTENCES, provide **relevant and specific** data/evidence from your lab experience.

- 1. What ingredient (or combination of ingredients) caused the fizzing?
- 2. Discuss several ways your items behaved in the solution.
- 3. Compare your data to other groups' data.

Comments: What did you learn by doing this lab? Provide relevant evidence.

Name	Period	Lab Report Due

<u>Lab #2- Observation Explosion!</u> Scoring guide

	On Time Neat	
1.	<u>Purpose</u> - Why <u>you</u> think we did this lab	
	10	
2.	<u>Hypothesis</u> - If, then, because	10
3.	<u>Observations</u>	
	- 3 labeled sketches with at least 3 detailed written descriptive pl	hrases per
	sketch.	30
4.	Gallery walk - with a series of sketches for each group, with oral con	clusion
		15
5.	Analysis Questions – Answer 3 Q's in complete sentences	
	with evidence from the lab experiences	15
6.	Comments- What did you learn by doing this lab?	
	Provide relevant evidence	20